Microscopic analysis of multipole susceptibility of actinide dioxides: A scenario of multipole ordering in AmO₂

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By evaluating multipole susceptibility of a seven-orbital impurity Anderson model with the use of a numerical renormalization group method, we discuss possible multipole states of actinide dioxides at low temperatures. In particular, here we point out a possible scenario for multipole ordering in americium dioxide. For Am^{4+} ion with five 5*f* electrons, it is considered that the ground state is Γ_7^- doublet and the first-excited state is Γ_8^- quartet, but we remark that the f^5 ground state is easily converted due to the competition between spin-orbit coupling and Coulomb interactions. Then, we find that the Γ_8^- quartet can be the ground state of AmO_2 even for the same crystalline electric-field potential. In the case of Γ_8^- quartet ground state, the numerical results suggest that high-order multipoles such as quadrupole and octupole can be relevant to AmO_2 .

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I. INTRODUCTION

Actinide dioxides with the fluorite structure of the space group Fm3m have been studied intensively for more than fifty years both from experimental and theoretical sides.^{1,2} A typical target material is UO₂ mainly due to its technological importance as a nuclear reactor fuel and a heterogeneous catalyst. As for theoretical research, a clear picture of the electronic structure in UO₂ has been obtained. In fact, neutron-scattering results show that UO₂ is a noncollinear antiferromagnet below 30.8 K.³ Detailed analysis of core photoemission spectra has suggested that UO₂ is an insulator of Mott-Hubbard type.⁴ The crystalline electric field (CEF) states of UO₂ have been also determined.⁵

Concerning NpO₂, over fifty years since 1953,⁶ it has been known to exhibit a mysterious low-temperature ordered phase.^{7,8} Several phenomenological works on the ordered phase have claimed a key role of octupole degree of freedom.^{9–13} The CEF states of NpO₂ have been determined by neutron-scattering experiment.¹⁴ Recently, the octupole order has been strongly supported by ¹⁷O-NMR experiment¹⁵ and by inelastic neutron-scattering study.¹⁶ In order to understand why such high-order multipole ordering appears, it is necessary to proceed to the research in a microscopic level. For this issue, it has been shown that octupole order actually characterizes the ground state of NpO₂ by the analysis of an *f*-electron model on the basis of a *j*-*j* coupling scheme on an fcc lattice.^{17,18}

On the other hand, PuO₂ is known to be a semiconductor with magnetic susceptibility which is almost independent of temperature up to 1000 K, since the CEF ground state is Γ_1^+ singlet and the first-excited state is Γ_4^+ triplet with the large excitation energy as 123 meV.^{19,20} Thus, from the viewpoint of magnetism, PuO₂ did not attract much attention.

Now let us turn our attention to AmO_2 . Probably due to the difficulty in the treatment of this material with high radioactivity, we cannot find lots of experimental results on AmO_2 . In 1969, Mössbauer isomer shift in AmO_2 was measured.²¹ After that, the magnetic susceptibility was measured and the peak was found around at 15 K.²² Naively thinking, such a peak seems to suggest the signal of antiferromagnetic ordering, while neutron-diffraction measurement did not detect antiferromagnetic order in agreement with the Mössbauer measurement.²³ This situation looks similar to that of NpO₂. Namely, multipole degree of freedom seems to be a key issue in AmO₂ to reconcile experimental results, as has been proposed for NpO₂ in the context of octupole ordering.

Here we note that the CEF ground state of AmO₂ was considered to be Γ_7^- doublet from the experimental results,^{22,24,25} but it could not bring higher multipoles. This is in sharp contrast to the case of NpO₂ with the confirmed CEF ground state of Γ_8^- quartet. Due to the CEF analysis of actinide dioxides, the Γ_7^- doublet ground state has been suggested.²⁶ Thus, it seems to be the mainstream in the research of actinide dioxides to clarify a mechanism which explains the disappearance of antiferromagnetic order for the Γ_7^- doublet ground state.

However, we believe that there still exists an alternative scenario on the basis of multipole ordering in AmO₂, when we recall the fact that the CEF ground state of the f^5 electron system is easily converted due to the competition between spin-orbit coupling and Coulomb interaction.²⁷ This point has been also discussed by the present author to propose a possible scenario which explains the change of the CEF ground state among Sm-based filled skutterudite compounds,²⁸ since trivalent Sm ion includes five 4f electrons. It has been experimentally²⁹⁻³² and theoretically²⁹⁻³³ shown that the angular momentum coupling of the 5f states of Am is situated between the LS and j-j coupling limits for many chemical situations, albeit closer to the j-j limit. Thus, a strong spin-orbit coupling is present in the 5f states of f^5 and f^6 configurations of Am that can compete with Coulomb interactions. Accordingly, it is believed to be meaningful to pursue a possibility of multipole ordering in AmO₂ with the Γ_8^- quartet ground state.

In this paper, we show that multipole order is possible in AmO₂, when we appropriately take into account both spinorbit coupling and Coulomb interaction in the *f*-electron terms for the CEF ground state. In the case of five *f* electrons such as Am⁴⁺ ion, the ground state is easily converted between Γ_7^- and Γ_8^- , when spin-orbit coupling and Coulomb interaction compete with each other. Then, the Γ_8^- quartet can be the ground state, even if the CEF potential is unchanged. In order to see what type of multipole is relevant, we evaluate the multipole susceptibility of the Anderson model by using a numerical renormalization group method. We find that higher-order multipoles are actually relevant to AmO₂ within the present calculation.

The organization of this paper is as follows. In Sec. II, we discuss the local *f*-electron state emerging from the competition among the Coulomb interaction and spin-orbit coupling under the CEF potential. In particular, the change of the f^5 ground state is explained in detail. In Sec. III, we show the model Hamiltonian. In order to discuss multipole properties, it is necessary to define the multipole operator. Here we explain the description of the multipole as spin-charge complex one-electron operator. Then, we briefly explain the numerical technique used in this paper. In Sec. IV, we show the results of the multipole state for the case of $n=2\sim 5$, where *n* is the local *f*-electron number. In particular, the results for n=5 are discussed in detail. Finally, the paper is summarized in Sec. V. Throughout this paper, we use such units as $\hbar = k_{\rm B} = 1$.

II. LOCAL f-ELECTRON STATE

Let us first discuss the local *f*-electron states of actinide ions. The local Hamiltonian should be composed of three parts as

$$H_{\rm loc} = H_{\rm so} + H_{\rm int} + H_{\rm CEF}.$$
 (1)

The first term denotes the spin-orbit coupling, given by

$$H_{\rm so} = \lambda \sum_{m,m'} \sum_{\sigma,\sigma'} \zeta_{m,\sigma,m',\sigma'} f^{\dagger}_{m\sigma} f_{m'\sigma'}, \qquad (2)$$

where $\sigma = +1$ (-1) for up (down) spin, $f_{m\sigma}$ is the annihilation operator for *f* electron with spin σ and *z*-component *m* of angular momentum $\ell = 3$, λ is the spin-orbit coupling, $\zeta_{m,\pm 1,m,\pm 1} = \pm m/2$, $\zeta_{m\pm 1,\mp 1,m,\pm 1} = \sqrt{12 - m(m\pm 1)}/2$, and zero for the other cases.

The second term indicates the Coulomb interaction among f electrons, expressed as

$$H_{\rm int} = \sum_{m_1 \sim m_4} \sum_{\sigma, \sigma'} I_{m_1 m_2, m_3 m_4} f^{\dagger}_{m_1 \sigma} f^{\dagger}_{m_2 \sigma'} f_{m_3 \sigma'} f_{m_4 \sigma}, \qquad (3)$$

where the Coulomb integral I_{m_1,m_2,m_3,m_4} is given by

$$I_{m_1,m_2,m_3,m_4} = \sum_{k=0}^{6} F^k c_k(m_1,m_4) c_k(m_2,m_3).$$
(4)

Here F^k is the radial integral for the *k*-th partial wave, called Slater integral or Slater-Condon parameter^{34,35} and c_k is the Gaunt coefficient.^{36,37} Note that the sum is limited by the Wigner-Eckart theorem to even values (k=0, 2, 4, and 6).

The third term is the CEF potential, given in the oneelectron potential form as

$$H_{\rm CEF} = \sum_{m,m'} \sum_{\sigma} B_{m,m'} f^{\dagger}_{m\sigma} f_{m'\sigma}, \qquad (5)$$

where $B_{m,m'}$ is the CEF potential. Since the fluorite structure belongs to $O_{\rm h}$ point group, $B_{m,m'}$ is given by using a couple of CEF parameters B_4^0 and B_6^0 for angular momentum $\ell=3$ as³⁸

$$B_{3,3} = B_{-3,-3} = 180B_4^0 + 180B_6^0,$$

$$B_{2,2} = B_{-2,-2} = -420B_4^0 - 1080B_6^0,$$

$$B_{1,1} = B_{-1,-1} = 60B_4^0 + 2700B_6^0,$$

$$B_{0,0} = 360B_4^0 - 3600B_6^0,$$

$$B_{3,-1} = B_{-3,1} = 60\sqrt{15}(B_4^0 - 21B_6^0),$$

$$B_{2,-2} = 300B_4^0 + 7560B_6^0.$$
 (6)

Note the relation of $B_{m,m'}=B_{m',m}$. Following the traditional notation,³⁹ we define

$$B_4^0 = Wx/F(4), \ B_6^0 = W(1 - |x|)/F(6),$$
 (7)

where x and the sign of W specify the CEF energy scheme, while |W| determines the energy scale for the CEF potential. Concerning nondimensional parameters, F(4) and F(6), we use F(4)=15 and F(6)=180 for $\ell=3.^{38}$

Here we briefly explain the parameters of the local Hamiltonian. Concerning Slater-Condon parameters, first we set $F^0=10$ eV by hand, since we are not interested in the determination of the absolute value of the ground-state energy. Others are determined so as to reproduce excitation spectra of U⁴⁺ ion with two 5*f* electrons.⁴⁰ Here we show only the results: $F^2=6.36$ eV, $F^4=5.63$ eV, and $F^6=4.13$ eV. As for spin-orbit coupling λ , we use the values of actinide atoms such as $\lambda=0.235$ (U), 0.272 (Np), 0.311 (Pu), and 0.351 eV (Am).

For the estimation of the CEF parameters, let us summarize the CEF energy levels of actinide dioxides. For UO₂, the ground state is Γ_5^+ triplet and the first-excited state is Γ_3^+ doublet with the excitation energy 150 meV.⁵ For NpO₂, the ground and first-excited states are, respectively, $\Gamma_8^{-(2)}$ and $\Gamma_8^{-(1)}$ quartets with the excitation energy 55 meV.¹⁴ For PuO₂, the ground state is Γ_1^+ singlet, while the first-excited state is Γ_4^+ triplet with the excitation energy 123 meV.^{19,20}

Now we set the CEF parameters for actinide dioxides in the present notation. For the purpose, first we estimate W and x so as to reproduce the CEF scheme of UO₂ by the f^2 electrons state. This is not a difficult task, since the CEF parameters are easily restricted from the experimental results. After that, among the values of W and x appropriate for UO₂, we further restrict the values of W and x which can also reproduce the results for NpO₂ and PuO₂. Note that the CEF states for f^3 and f^4 states are almost reproduced by using the parameters of f^2 electron state. Since the CEF term is just given by the one-electron potential, the CEF effect is not so drastically changed among the materials with the same crystal structure, even though the f-electron number is different.



FIG. 1. CEF energy level schemes for tetravalent actinide ions. Parameters used here are explained in the main text.

After some calculations, we determine W=-10.5 meV and x=0.62. The results for the CEF level schemes are summarized in Fig. 1. It is observed that the CEF states of UO₂, NpO₂, and PuO₂ are well reproduced. Note here that for the same CEF parameters, the ground state for Am⁴⁺ is found to be Γ_7^- doublet and the excited state is Γ_8^- quartet with the excitation energy of about 50 meV. This is consistent with the previous theoretical results on the CEF states of AmO₂ obtained by more detailed calculations.²⁶

Naively thinking, there occurs ordering of magnetic moment originating from Γ_7^- ground state for AmO₂, but it seems to be contradict with neutron-diffraction study. In order to resolve such contradiction, there are two ways: one is to consider a mechanism which explains the disappearance of magnetic moment even in the Γ_7^- ground state. Another way is to reconsider the local *f*-electron term by focusing on spin-orbit coupling and Coulomb interaction. Here we propose an alternative scenario on the basis of the second direction.

Thus far, we have simply assumed that Coulomb interaction is not changed among different actinide ions, but in actuality, they may be changed. Concerning spin-orbit coupling, we have used the values in actinide atom, but it may be also changed in the ionic states. Then, we point out that for f^5 systems, the CEF states are sensitively changed by the competition between Coulomb interaction and spin-orbit coupling in comparison with other values of local *f*-electron number.^{27,28}

In order to understand that the effect of Coulomb interaction and spin-orbit coupling appears in the CEF parameters, here we express the CEF parameter B_4^0 by using the so-called Stevens' factor as

$$B_4^0 = A_4 \langle r^4 \rangle \beta_J^{(n)}, \tag{8}$$

where A_k is the parameter depending on materials, $\langle r^k \rangle$ denotes the radial average of local *f*-electron wave function, *n* denotes the local *f*-electron number, *J* is the total angular momentum of the ground-state multiplet, and $\beta_J^{(n)}$ indicates the Stevens' factor, which is one of coefficients appearing in the method of Stevens' operator equivalent.⁴¹

For the case of n=5, it is well known that the ground-state multiplet is characterized by J=5/2. After lengthy calculations, for n=5 and J=5/2, we can obtain that $\beta_{5/2}^{(5)} = (13/21)\beta_3$ in the *LS* coupling scheme and $\beta_{5/2}^{(5)} = -(11/7)\beta_3$ in the *j*-*j* coupling scheme,²⁷ where β_3 denotes the Stevens factor for $\ell=3$, given by $\beta_3=2/495$. It should be



FIG. 2. (Color online) The CEF energy for n=5 vs r. The meaning of r is explained in the main text.

noted that the sign of $\beta_J^{(n)}$ is changed between the *LS* and *j*-*j* coupling schemes, suggesting that the ground state is converted, when Coulomb interaction and/or spin-orbit coupling are changed.

The modification of Coulomb interaction and spin-orbit coupling is closely related to the picture for multi *f*-electron state. The CEF level schemes in Fig. 1 is qualitatively understood by a *j*-*j* coupling scheme. Namely, by assuming that the effective Hund's rule coupling is smaller than the CEF level splitting, we simply accommodate plural numbers of *f* electrons in the levels of Γ_8^- ground and Γ_7^- excited states. Then, we can easily reproduce all the CEF level schemes of tetravalent actinide ions. However, the conversion of the CEF ground state for n=5 indicates that the actual situation should be slightly shifted to the side of the *LS* coupling scheme. We note that due to such a shift, the CEF states for n=2, 3, and 4 are not qualitatively changed, while in the case of n=5, the ground state is converted.

Here we emphasize that such a ground-state conversion occurs in the region of realistic values of Coulomb interaction and spin-orbit coupling, as has been pointed out in the discussion of the CEF states of Sm-based filled skutterudites.²⁸ In fact, as mentioned in Sec. I, it has been shown that the angular momentum coupling of the 5*f* states of Am is situated between the *LS* and *j*-*j* coupling limits, but it is rather closer to the *j*-*j* coupling limit.^{29–33} Thus, it seems to be reasonable to change slightly spin-orbit coupling and/or Coulomb interaction in the case of tetravalent Am ion.

If we resort to first-principles calculation, we may determine correctly F^j and λ , but it is out of the scope of the present paper. Here we simply introduce an artificial parameter *r* to control Coulomb interactions and spin-orbit coupling as $F^k \rightarrow rF^k$ and $\lambda \rightarrow \lambda/r$ with $r \ge 1$.

In Fig. 2, we depict the energies of $H_{\rm loc}$ as functions of r for n=5. We find that the conversion of the ground state occurs around at $r \approx 1.3$. Note that in other values of n, the ground states are not changed. Thus, we can obtain the Γ_8^- quartet ground state for Am⁴⁺. In the following sections, we will discuss how the multipole states actually appear when we change the value of r for Am ion.

III. MODEL AND METHOD

A. Anderson model

Now we include the hybridization between localized and conduction electrons. The Hamiltonian is the Anderson model, given by

$$H = \sum_{\mathbf{k},\sigma} \varepsilon_{\mathbf{k}\sigma}^{\dagger} c_{\mathbf{k}\sigma} + \sum_{\mathbf{k},\sigma,m} \left(V_m c_{\mathbf{k}\sigma}^{\dagger} f_{m\sigma} + \text{h.c.} \right) + H_{\text{loc}}, \quad (9)$$

where $\varepsilon_{\mathbf{k}}$ denotes conduction electron dispersion, $c_{\mathbf{k}\sigma}$ indicates the annihilation operator for conduction electron with momentum \mathbf{k} and spin σ , and V_m is the hybridization between conduction and f electrons.

Note that we consider only a_{μ} single conduction band with xyz symmetry composed of oxygen 2p electrons. Since oxygen ions surrounding actinide ions are located in the [1, 1, 1] direction, there should exist a conduction band composed of 2p electrons with xyz symmetry. This picture seems to be consistent with band-structure calculation,⁴² but the ignorance of t_{1u} and t_{2u} bands is just assumption. Here we note that the hybridization occurs between the states with the same symmetry of local *f*-electron state. Since the a_{μ} conduction band has xyz symmetry, we set $V_2 = -V_{-2} = V$ and zero for other *m*. Hereafter, a half of the bandwidth of $a_{\rm u}$ conduction band D is set as the energy unit, i.e., D=1 eV. We fix V as V/D=0.05 throughout this paper. Note that in order to adjust the local f-electron number n, we appropriately change the chemical potential in the actual calculation, although we do not explicitly show such a term.

B. Multipole operator

In order to discuss multipole properties, it is necessary to define the multipole operator \hat{X} for f electron.^{43–48} In general, \hat{X} is expressed as

$$\hat{X} = \sum_{k,\gamma} p_{\gamma}^{(k)} \hat{T}_{\gamma}^{(k)},$$
(10)

where k is a rank of multipole, γ is a label to express $O_{\rm h}$ irreducible representation, and $\hat{T}_{\gamma}^{(k)}$ is cubic tensor operator, given by $\hat{T}_{\gamma}^{(k)} = \sum_{q} G_{\gamma,q}^{(k)} \hat{T}_{q}^{(k)}$. Here an integer q runs between -k and k, $\hat{T}_{q}^{(k)}$ is spherical tensor operator, and $G_{\gamma,q}^{(k)}$ is the transformation matrix between spherical and cubic harmonics. We determine $p_{\gamma}^{(k)}$ later.

In order to obtain explicit expression of the spherical tensor operator $\hat{T}_q^{(k)}$, it is convenient to convert the *f*-electron basis from (m, σ) to (j, μ) , where *j* is the total angular momentum and μ is the *z* component of *j*. When we define $f_{j\mu}$ as the annihilation operator for *f* electron labeled by *j* and μ , we obtain $\hat{T}_a^{(k)}$ in the second-quantized form as

$$\hat{T}_{q}^{(k)} = \sum_{j,\mu,\mu'} T_{j;\mu,\mu'}^{(k,q)} f_{j\mu}^{\dagger} f_{j\mu'}.$$
(11)

Note that there are no components between different values of *j*, since the matrix for total angular momentum is block-diagonalized in the basis of (j, μ) . The matrix element of $T_{j;\mu,\mu'}^{(k,q)}$ is calculated by the Wigner-Eckart theorem as

$$T_{j;\mu,\mu'}^{(k,q)} = \frac{\langle j || T^{(k)} || j \rangle}{\sqrt{2j+1}} \langle j \mu | j \mu' kq \rangle, \tag{12}$$

where $\langle j\mu | j\mu' kq \rangle$ indicates the Clebsch-Gordan coefficient and $\langle j||T^{(k)}||j\rangle$ denotes the reduced matrix element for spherical tensor operator, given by $\langle j||T^{(k)}||j\rangle$ $=\sqrt{(2j+k+1)!/(2j-k)!/2^k}$. Note that $k \le 2j$ and the highest rank of *f*-electron multipole is seven.

Let us now determine the coefficient $p_{\gamma}^{(k)}$. In order to discuss the multipole state, it is necessary to evaluate the multipole susceptibility in the linear response theory. However, multipoles belonging to the same symmetry are mixed in general, even if the rank is different. In addition, multipoles are also mixed due to the CEF effect. Thus, we determine $p_{\gamma}^{(k)}$ by the normalized eigenstate of susceptibility matrix

$$\chi_{k\gamma,k'\gamma'} = \frac{1}{Z} \sum_{i,j} \frac{e^{-E_i/T} - e^{-E_j/T}}{E_j - E_i} \langle i | [\hat{T}_{\gamma}^{(k)} - \rho_{\gamma}^{(k)}] | j \rangle \\ \times \langle j | [\hat{T}_{\gamma'}^{(k')} - \rho_{\gamma'}^{(k')}] | i \rangle,$$
(13)

where E_i is the eigenenergy for the *i*-th eigenstate $|i\rangle$ of H, T is a temperature, $\rho_{\gamma}^{(k)} = \sum_i e^{-E_i/T} \langle i | \hat{T}_{i\gamma}^{(k)} | i \rangle / Z$, and Z is the partition function given by $Z = \sum_i e^{-E_i/T}$. Note that the multipole susceptibility is given by the eigenvalue of the susceptibility matrix.

C. Method

In order to evaluate multipole susceptibility of the Anderson model, we employ a numerical renormalization group (NRG) method.⁴⁹ in which momentum space is logarithmically discretized to include efficiently the conduction electrons near the Fermi energy and the conduction electron states are characterized by "shell" labeled by *N*. The shell of N=0 denotes an impurity site described by the local Hamiltonian.

In the NRG method, we transform the Hamiltonian into the recursion form as

$$H_{N+1} = \sqrt{\Lambda} H_N + t_N \sum_{\sigma} \left(c_{N\sigma}^{\dagger} c_{N+1\sigma} + c_{N+1\sigma}^{\dagger} c_{N\sigma} \right), \quad (14)$$

where Λ denotes a parameter for logarithmic discretization, $c_{N\sigma}$ indicates the annihilation operator of conduction electron in the *N* shell, and t_N is the hopping of electron between *N* and (N+1) shells, expressed by

$$t_N = \frac{(1 + \Lambda^{-1})(1 - \Lambda^{-N-1})}{2\sqrt{(1 - \Lambda^{-2N-1})(1 - \Lambda^{-2N-3})}}.$$
 (15)

The initial term H_0 is given by

$$H_0 = \Lambda^{-1/2} \left[H_{\text{loc}} + \sum_{\sigma} V(c_{0\sigma}^{\dagger} f_{c\sigma} + f_{c\sigma}^{\dagger} c_{0\sigma}) \right].$$
(16)

Each component of multipole susceptibility matrix Eq. (13) is evaluated by using the renormalized state. Then, the multipole state is defined by the eigenstates of Eq. (13). We note that the temperature *T* is defined as $T = \Lambda^{-(N-1)/2}$ in the NRG calculation, where *N* is the number of the renormalization step. Due to the limitation of computer resources, we keep only *M* low-energy states. In this paper, we set $\Lambda = 5$ and M = 3000.

IV. RESULTS

Now we discuss the multipole state of the Anderson model Eq. (9). In Figs. 3(a)-3(c), we show $T\chi_{\gamma}$ vs T for n



FIG. 3. (Color online) $T\chi_{\gamma}$ vs temperature T for (a) UO₂, (b) NpO₂, and (c) PuO₂.

=2, 3, and 4, where χ_{γ} is the eigenvalue of the multipole susceptibility. We use the same values of the parameters in the Hamiltonian for n=2, 3, and 4. The values of the spinorbit coupling are $\lambda=0.235$, 0.272, and 0.311 eV for n=2, 3, and 4, respectively. The eigenstates are classified by irreducible representation of $O_{\rm h}$ point group. Here we use shorthand notations such as "3g" and "5u," which denote Γ_3^+ and Γ_5^- , respectively, in the Bethe notation. Note that "1u" does not appear among multipoles up to rank seven.

For $n=2(UO_2)$, we find the optimized state is labeled by 4u, which is mainly composed of dipole (about 92%). The secondary components are quadrupoles (5 and 3g), but χ_{5g} and χ_{3g} is smaller in one order in comparison with χ_{4u} . For $n=3(NpO_2)$, as easily deduced from the Γ_8^- quartet ground state, we find varieties of multipoles. Among them, the primary component is 4u, which is mainly composed of dipole (about 96%). The secondary one is 2u octupole. In the third group, another 4 and 5u are almost degenerate. We note that this 4u is composed of higher multipole component and the 5u is mainly composed of octupole. The fourth component is 3g quadrupole. We emphasize the existence of octupoles (2 and 5u) with significant eigenvalues.

In the present calculation, we cannot determine the kind of multipole ordering in actual systems. However, the multi-



FIG. 4. (Color online) (a) $T\chi_{\gamma}$ vs temperature *T* for n=5. The parameters are the same as those in Fig. 2, except for the spin-orbit coupling. (b) $T\chi_{\gamma}$ vs temperature *T* for n=5 and r=1.5 with enhanced Coulomb interaction and reduced spin-orbit coupling.

poles which remain at low temperatures are the candidates which will order in the actual system. In the case of NpO₂, it has been gradually revealed that triple- \mathbf{q} order of 5u octupole can naturally reconcile several kinds of experiments. The 5u octupole is actually included in the multipoles in the present calculations, although it is not dominant component.

For n=4 (PuO₂), we do not find any significant multipole component, as easily understood from the Γ_1^+ singlet ground state which is well separated from the magnetic excited state. In this sense, from the viewpoint of magnetism, this case does not attract much attention.

Next we move on to the case of n=5, corresponding to AmO₂. In Fig. 4(a), we show the results for n=5 by using the parameters for r=1. Namely, the Coulomb interactions are the same as those in Figs. 3(a)-3(c). The spin-orbit coupling is set as $\lambda=0.351$ eV, which is the value for Am atom. In this case, since the ground state is Γ_7^- doublet, the component which remains in the low-temperature region is 4u, which is composed of dipole (about 25%) and octupole (about 75%). We note that the octupole component is significantly large in comparison with the 4u states of UO₂ and NpO₂. In any case, as we have expected, we find only magnetic 4u moment originating from Γ_7^- ground state for AmO₂.

Then, we increase the value of r in order to move to the side of the *LS* coupling scheme. In Fig. 4(b), we plot $T\chi_{\gamma}$ of the multipole susceptibility for n=5 and r=1.5 with the Γ_8^- ground state (see Fig. 2). In this case, we find that the primary component is 3g, which is mainly composed of quadrupole. The secondary component is 4u and we also find 5g, 4u, and 5u components with smaller eigenvalues. We note that some multipoles of AmO₂ are the same as those in NpO₂, except for 2u and 5g, although the corresponding eigenvalues are different.

Now we provide a comment on the value of r, which is introduced so as to increase the effect of Coulomb interactions and decrease the magnitude of spin-orbit coupling. Note, however, that the angular momentum coupling of the 5f states of Am is nowhere near as close to the *LS* coupling limit as for Cm.^{31,33} Thus, we should not entirely or strongly suppress the spin-orbit coupling in the 5f states of Am. It is acceptable that we change slightly spin-orbit coupling and Coulomb interaction. Accordingly, when the value of r is increased so as to move toward the side of the *LS* coupling scheme, we should pay due attention *not* to suppress spinorbit coupling too much.

V. DISCUSSION AND SUMMARY

In this paper, we have discussed the multipole state in the low-temperature region, by analyzing the seven-orbital impurity Anderson model with the use of the NRG method. We have found the multipole state for n=2, 3, and 4, which are not in contradiction to the phases observed in UO₂, NpO₂, and PuO₂, respectively. Note here that we determine the candidates which will order in the actual periodic system at low temperatures.

For the case of n=5, when we use the same parameters as those for $n=2\sim4$, we have suggested the phase dominated by magnetic moment. However, if we change slightly Coulomb interactions and spin-orbit coupling, we have found the Γ_8^- quartet ground state for AmO₂. In this situation, we have shown that the low-temperature phase can contain multipoles such as quadrupole and octupole. The parameters are artificially introduced here, but our purpose is to point out a possibility of the Γ_8^- quartet ground state due to the competition between Coulomb interactions and spin-orbit coupling.

Unfortunately, we cannot determine the kind of multipole order only from the present calculation, but on the basis of the same crystal structure, it is plausible that 5u octupole order also appears in AmO_2 . On the other hand, it may be possible to exploit other scenarios, e.g., quadrupole ordering, which were invented for understanding of NpO_2 . In any cases, the combination of phenomenological theory and microscopic experiment will be useful to finalize the kind of multipole which orders at low temperatures in AmO_2 .

Experimentally it has been considered that the CEF ground state of AmO_2 is Γ_7^- ground state. However, we believe that it is still meaningful to examine the experimental results on the basis of the Γ_8^- quartet ground state, although it may be difficult to perform the microscopic experiments of AmO_2 .

Finally, let us provide a comment on the simplification of the model. In this paper, since we have considered only single conduction band, there exists residual entropy in the results. In actuality, it should be finally released when we consider t_{1u} and t_{2u} conduction bands. This point is also related to the relevant multipole moment when we consider the ordered state in the periodic systems. In this sense, the present results are qualitative, but they include the actual multipole which forms ordered state.

In summary, we have discussed the multipole state of actinide dioxides due to the evaluation of the multipole susceptibility of the Anderson model. When Coulomb interaction and spin-orbit coupling have been appropriately changed, it has been found that multipoles including quadrupole and octupole are relevant to AmO_2 . It is believed that multipole ordering can be detected in AmO_2 in future experiments.

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